The Effect of Aging on the Strength of Newly Developed Corrosion Resistance Resistant Incoloy Alloys 945 and 945X from Aa Microstructural Perspective

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# Abstract

It has been found The results of this study show that Incoloy 945X has is 19.36% more hardness harder and 10.91% more strengthstronger than Incology 945. The Of particular interest is the morphology of precipitates morphology responsible for precipitation hardening in both alloys is of particular interest in current investigation. To identify and understand the physical reason for this, mechanisms that lead to these results, we undertook an extensive microstructural study is performed wherein which  $\gamma$  size, shape and elemental constitution iswas investigated and correlated to the aging conditions. The-The main strengthening phase for both alloys is identified by transmission electron microscopy studies reveal only one to be the main population of  $\gamma$ ' precipitates sized, which are between 10 and 25 nm as the main strengthening phase for both alloys.in size. Scanning transmission electron microscopy line scans across the  $\gamma'$  precipitate exhibit reveal more Nb and Ti deflection in Incoloy 945X than in Incoloy 945. This is further supported by the results of transmission electron microscopy findings, in which  $\gamma$ in 945X appear bright (because of Nb) with faceted growth, resulting in a higher value of 0.2% yield strength (1100 MPa) and hardness (430 HV5) than the Incoloy 945. Coarsening-For each aging temperature, the coarsening kinetics of the  $\gamma$  phase were was simulated for each aging temperature by using JMatPro. The coarsening simulations were found to be in good agreement, and the simulation <u>results were consistent</u> with the experimental results for the mean diameter of the  $\gamma$ ' precipitates in each alloy.

**Keywords:** 945; 945X; Scanning electron microscopy; Transmission electron microscopy;  $\gamma$ '; Coarsening simulations, precipitation, strengthening-

#### 1. Introduction

The high-strength Incoloy alloy 945 was developed in 2008 by the Special Metals Corporation for the oil and gas industry, as it and is more suitable than the traditional Inconel alloy 718 for deep-drilling applications that involve deep drilling. This alloy. Incoloy 945 is a precipitation-hardened and corrosion \_resistant Ni-Fe-Cr alloy. Although its-it contains less nickel content-than Inconel 718 (53 wt%), this alloy provides excellent ductility, impact strength<sub>7</sub> and resistance to stress corrosion cracking. In 2011, a higher—strength version of the alloy, called the <u>"Incoloy 945X2945X"</u> (54 wt% nickel content),

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**Commented [SC ED4]:** Please provide full postal address of all the affiliations of the authors you have mentioned above. An example of the format is given below: <sup>1</sup>First Department, First University, Address, City, Country Name Also, mention the email address of each author, if available.

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was introduced. These Incoloy alloys can be used in high-pressure, high-temperature (HPHT) applications; and are approved for use under NACE MR0175/ISO 15156-3 up to levels V and VI [1].

As is normal with any Mnickel-based superalloy, the matrix consists of a face-centred cubic (FCC) fcc) structure called gamma ( $\gamma$ ). The solid solution strengthening elements of gamma are chromium, iron, molybdenum, titanium, aluminium, and cobalt. The main strengthening precipitates,  $\gamma'$  (Ni<sub>3</sub>(Al, Ti, Nb) and  $\gamma''_{(Ni_3Nb)}$  havinge L1<sub>2</sub> and DO<sub>22</sub> crystal structures, respectively, and have been the subject of numerous investigations [2-7]. As reported by Hagel and Beattie [8], the  $\gamma'$  phase is coherent with the gamma matrix, with lattice ( $\gamma$ ) misfits of 0.1 %, 0.5-%-1.0-%-% and greater than 1.25 % associated with the spherical particles, cubes<sub>7</sub> and plates, respectively. The  $\gamma'$  phase is an A<sub>3</sub>B phase, in which the element substituting A is Ni, whilewhereas Co and B are replaced by Ti, Al<sub>7</sub> and Nb. Hume-Rothery [9], [36,37-11] stated that Ni<sub>3</sub>Al is the most stable  $\gamma'$  phase, followed by Ni<sub>3</sub>Ti, Ni<sub>3</sub>Nb<sub>7</sub> and Ni<sub>3</sub>Ta, in the order of decreasing stability.

A nickel content above 42 wt.% is typically required for resistance to aqueous stress corrosion cracking [1012]. Molybdenum and chromium in the alloy provide resistance to pitting corrosion in reducing acid and alkali solutions [++13,+214]. Niobium, titanium, and aluminium are the primary precipitation-strengthening components [1315]. Although titanium increases the strength and hardness of the alloy, an excessive amount of *ittitanium* is known to degrade the microstructural stability [4416]. Aluminium increases the volume fraction of the  $\gamma'$  phase [4517], which is expected to further increase hardness; however, a high aluminium content reduces the lattice mismatch/ and coherency strain, resulting in a net decrease in hardness. The metastable  $\gamma''$  phase [5], [7], [4618-+820] has a body-centred tetragonal structure (BCT), and usually precipitates at temperatures less than 650 °C. The nickel alloys normally exhibit undergoes  $\gamma$ ? precipitation of  $\gamma$ ? under high coherency strains [1315]. These are fine nanoThe precipitates are fine nanoparticles, like  $\gamma'$ , with high hardening and strength, having a and with length approximately six times greater than their width. This metastable phase tends to be replaced by a stable delta Ni<sub>3</sub>Nb ( $\delta$ ) phase when exposed for extended periods to temperatures above 650 °C-for extended periods [3521]. The presence of intergranular  $\delta$  platelets [1922] with an orthorhombic crystal structure results in a loss of hardness and-ductility [2023].

This study seeks to verify whether Incoloy alloys 945 and 945X are known to resist the formation of intergranular phases, and this study seeks to verify this as is suspected. This resistance could be due to a lower Nb content of Nb in the actual alloy composition of the alloys, as compared towith the Nb content in the parent alloy, Inconel 718. Typically, the nickel-based alloys have low carbon content is low in Nickel based alloys. Carbon tends to react with refractory elements to form MC-type carbides that reside at the grain boundaries and within grains, with the 'M' mostlymost often being Nb. Furthermore, Ti helps to form TiN; both the MC carbides and TiN are beneficial forhelp

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W. Hume-Rothery, G. W. Mabbott, and K. M. Channel-Evans, "The freezing points, melting points, and solid solubility limits of the alloys of silver and copper with the elements of the B Sub-Groups", Philos. Trans. R. Soc. London, Ser. A. 233, 1934, 1-97.

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strengthen the grain boundary strengthening, inhibition of boundaries, inhibit grain growth and reduction of reduce grain sliding, despite their detrimental properties of reduced ductility and toughness. Few studies [2+24,2+25] have been carried outfocussed on 945 and the relatively new 945X alloys. The relationship between the microstructure and mechanical properties for these alloys haves not been characterised as fully<sub>7</sub> as compared to has that of other Nnickel-based alloys. The aim of the current study is to investigate the microstructure of the 945 and 945X alloys under isothermal solution treatment and aging- and analyse their effects on how it affects the mechanical strength. The reason for which of the alloy.

#### 2. Experimental Methods

#### 2.1. Experiment Materials

Incoloy alloys 945 ( $140 \times 110 \times 24 \text{ mm}^3$ ) and 945X ( $250 \times 250 \times 360 \text{ mm}$ ), in its as received condition, mm<sup>3</sup>) were supplied by Titanium Engineers, Inc and C. H. Robinson Europe BV, respectively, and used as received. The as received alloys presented had the microstructures of the type of microstructure typically used in oil and gas applications. The alloys in the as received condition and were termed called 'as 'As Received'received' (AR) alloys and details, Details of the AR heat treatment [1416] areappear in Table 21. The Table 12 shows the results of the optical emission spectroscopy (OES) studies carried out on applied to thin off-cuts from each alloy, are shown in Table 1.

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Specimen	Alloy	Dimensions	Heat Treatment	
ST	945/945X	14 <del>x ×</del> 10	1038 ⊜ <u></u> C 1 h WQ	Commented [EN QA23]: I have edited the degree symbol
		mm <sup>2</sup>		"o".
STA	945/945X	14 <del>x_</del> 10	1038 °C 1 h WQ then 718 °C 8 h FC + 621	
		<u>mm<sup>2</sup></u>	°•C 8 h AC	
			1010-1066 $^{\circ}\mathrm{C}$ for 0.5-4 h, WQ to RT	
AR	945/945X	14 <del>x_</del> 10	704-732 °C for 6-8 h, FC at 0.433-	
		mm <sup>2</sup>	0.933 °C/min to 607-635 °C, then hold for	
			6-8 h, AC to RT	
NS	945X	14 <del>x−</del> <u>×</u> 10	1250 °C 1 h WQ then 718 °C 8 h FC + 621	
		mm <u>2</u>	°⊖C 8 h AC	

The 945 and 945X alloys were subjected to thermal heat treatments under the conditions described in Table 2. The samples were ST in a preheated furnace to 1038 °C for 1 h and then quenched with water to RT 20 °C at a cooling rate of 50 °C/min. The ST samples were prepared for microstructural investigations, while the remaining samples were STA in a preheated furnace to 718 °C and aged at this temperature for 8 h.

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Table 21. Solution annealing and aging temperature of the 945 and 945X specimens.

Alloy	wt.	Si	Mn	Cr	Мо	Fe	Al	Cu	Nb	Ti	Ni
	% C										
945	0.01	0.1	0.07	20.5	3.2	22.2	0.17	1.95	3.11	1.55	47.04
945X	0.01	0.08	0.04	20.2	3.2	14.4	0.15	2.09	4.14	1.57	53.9

 Table 12 Elemental composition of as\_received 945 and 945X alloys.

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### 2.2. Microstructural studies

Metallographic samples for <u>lightoptical</u> microscopy (LM)-and scanning electron microscopy (SEM)- $\frac{1}{3}$  XL30 ESEM with Oxford ISIS 310 EDX facilities-(\_operated at 10 kV $\frac{1}{5}$  spot size  $5\frac{1}{5}$  working distance 5; Backscatter Electron, backscatter electron mode) were prepared by using standard mechanical polishing procedures. The specimens to be observed-were ground by using silicon carbide papers with P240–P2500 grit sizes and polished by using the abrasive cloths in a-descending order of grit sizes from 6 µm to ~0.25 µm. Kalling's reagent etchant (100 ml HCl, 100 ml ethanol, 5 g CuCl<sub>2</sub>) and colloidal silica (0.06 µm) were used to reveal the microstructure (i.e. the grain boundaries and metal carbides-). Transmission electron microscopy studies were carried outwas done with thea JEOL 2100 operated at an accelerationg voltage of 200 kV (beam current ~102 µA) with a spot size of 1 and an alpha angle of 2°. Thin foils prepared from each sample were mechanically ground down-to-a thickness less than 100 µm thick by using the silicon carbide papers (240–1200 grit). 3-mm

### 2.3. Mechanical Studies

To investigate the strength of the  $\gamma'$  precipitates, each sample was subjected to Vickers micro-hardness (HV) measurements were carried out for each sample, using done with a Rockwell Indentation Hardness Tester by (Tecquipment-ltd, to obtain insights on the strength of the  $\gamma'$  precipitates., Ltd.). These measurements were performed by using a diamond pyramid indenter with a force of 5 kgf (HV5). The set of values were recorded while moving from one end of the sample to the other (unit:units of mm). The To prevent overlap of the strain fields, the distance between each indentation was approximately > $\geq$ 3d, where d is the indent diameter (in either inthe x or y direction), to prevent overlap

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of the strain fields.). Tensile specimens from each alloy were machined as perto the following dimensions: gage length of 22.5 mm, diameter of 6.35 mm, and width of the grip section of 12.5 mm, with M8 threaded ends. The specimens were tested by using an Instron tensile machine at a standard displacement rate (SDR) of 2.22  $\# \times 10^{-4}$  s<sup>-1</sup>. SEM with energy dispersive x ray spectrometry (\_EDS) was conducted used to determine the compositions of the various phases. Seanning transmission electron microscospy (\_and\_STEM) was used to identify and obtain the chemical constituents of the nano precipitates. To simulate the precipitation kinetics of the  $\gamma$ ' precipitate in the 945 and 945X alloys, a we used the JMatPro thermo-kinetic software package known as JMatPro (version 8.04) was used; and compared the results from the simulation and experiments were then compared. The with those from experiment. These simulations helped provide provided insights oninto the size of the  $\gamma$ ' precipitates for, which could affect the mechanical properties of the alloy could be affected.

## 2.3. Results and Discussion

#### 3.1. Grain size after heat treatment

Light-Optical-microscopy images were obtained for the ST, STA<sub>7</sub> and AR samples of each alloy. Fig-ure 1-(a) isshows a typical image of alloys 945 and 945X, containingwhich contain a  $\gamma$  matrix with grains and grain boundaries. The nano scaleIn the subsequent sections, the nanoscale  $\gamma'$  phase in the  $\gamma$  matrix is examined for each alloy sample-in the subsequent sections. Initially, an uneven microstructure was obtained by swabbing the samples with a cotton bud containing kKalling's reagent (concentrated and dilute solutions); some grains were etched whilewhereas others remainsed under-etched with no clear distinction between the grain boundaries. By using colloidal silica (0.04  $\mu$ m in size) as the final polishing suspension, the grains were not only polished, but also slightly etched, thusthereby revealing the grain boundaries. Grain sizes were evaluated by using the intercept method, by and considering 10 lightoptical micrographs of each sample. No significant variation in the grain size was observed, and an average grain size of 110  $\mu$ m was recorded for appeared, and each standard alloy specimen, had an average grain size of 110  $\mu$ m. However, due to ST at 1250 °C, the grain size in the 945X NS sample had an increased grain size in the range ofto 140–150  $\mu$ m, and exhibited an intergranular phase after double aging, as shown in [see Fig. 1-(b)-)].

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The electron diffraction pattern (Fig. 6, left) for the 945 ST sample is more associated with a fcc  $\gamma$  lattice; however, due to the presence of small-sized  $\gamma$ ' precipitates, they are more coherent with the matrix and not easily distinguishable. In contrast, for the 945X ST sample (Fig. 6, right), upon higher magnification (([Figs. 6(b) and 6(c)),], the  $\gamma$ ' precipitates can beare clearly distinguished from the  $\gamma$  matrix. This is further evident from upported by the electron diffraction spots offrom the  $\gamma$  matrix

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and  $\gamma$ ' precipitates. These very small precipitates give rise to secondary diffraction spots in the selected-area diffraction patterns, and are primarily associated with the presence of the  $\gamma$ ' phase in the matrix.

**Fig. 4.** TEM images of 945 (left) and 945X (right) AR sample. Left: (a) Dark—field imaging— Overview of  $\gamma$ ' precipitates (appearing–bright)—which\_spots) that are distributed homogeneously throughout the  $\gamma$  matrix with selected—area diffraction pattern (SADP)—corresponding to ( $\overline{110}$ )  $\gamma'$ phase. (b) Bright—field mode showing contrast image of  $\gamma'$  and  $\gamma$  matrix with a faint line around precipitates distinguishing  $\gamma'$  precipitates from  $\gamma$  phase and (c) at a higher magnification, clearly showing  $\gamma'$  appearing black in a spherical shape arboundt [4–8 nm in diameter and in a compact disc shaped ranging between shape about 10–20 nm in diameter. Right: 945X AR: (a) shows (0 $\overline{10}$ )  $\gamma'$ precipitates as grey precipitates in dark—field mode with the  $\gamma$  matrix appearing as dark grey. Panels (b) and (c) correspond+ to higher magnification, clearly showing  $\gamma'$  inwith spheroid shape (10–15 nm) and coalescence of  $\gamma'$  into aggregates 10–25 nm in size.

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Figure 7 shows the size distribution of  $\gamma'$  in the AR and STA samples-are presented in Fig. 7. The graph clearly indicates a peak position where the maximum number of  $\gamma'$  precipitates are found in any sample. The distribution is not a normal distribution, as there are because some secondary peaks appear at 0–5–nm, 10–15 nm, and 15–20 nm for each alloy. The error bars associated with the distribution data isare ±0.5 nm for each reading. The graph for the 945 AR and STA samples shows that the mean size range of  $\gamma'$  is between ranges from 6–to 8 nm, while whereas that for the 945X AR and STA samples shows that the mean  $\gamma'$  size range increases to ranges from 10– to 12 nm. It is important to noteNote that, for each alloy, the curves for the AR samples nearly overlap with the curves for the STA samples, for each alloy. It can be assumed that the AR samples of each alloy have undergone the same heat treatment as the STA samples, A mean size difference of ~4 nm between the 945 STA and 945X STA samples was recorded is found for  $\gamma'$ . It is important to consider hereNote that itthis is not the mean, but the average  $\gamma'$  size of each alloy, which allows us to predict their cutting or bowing phenomena under service stress and cathodic protection.

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As discussed earlier, the amount of Nb that is present in 945 and 945X is arboundt 3.11 wt% and 4.14 wt%, as compared towhich differs from its parent alloys Inconel 718 (5.06 wt%) and Incoloy 925 (0 wt%), respectively. This reduction in the Nb content aids in resistinghelps to resist the

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precipitation of metastable  $\gamma$ '' (Ni<sub>3</sub>Nb) and the stable  $\delta$  phase. This change in chemistry results in the precipitation of only the  $\gamma$ ' phase at the nano scale,nanoscale and is evident from the TEM images in Fig. 4-1.6 (see section 3.3.1). HThis conclusion is further supported by the findings of Slama and Abdellaoui [2428], who reported that  $\gamma$ ' appears before  $\gamma$ '' during isothermal aging at 680 °C. The authors

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**This increases further with the increase in the size of \gamma'. InFor** the 945X samples, the  $\gamma$ ' precipitates appear bright (Figs. 4 and 1.5), which could be attributed to the presence of higher-atomic-number elements <u>likesuch as</u> Nb, which diffuse at the edges of the  $\gamma$ ' precipitates to minimise the elastic strain caused by the lattice misfit between the  $\gamma$ ' phase and  $\gamma$  matrixces [2832,2933].

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The hardness values for the 945 and 945X ST samples are not surprising, essince incomplete chemical ordering within the  $\gamma'$  precipitates and a small size contributes partially to the strength (see Fig. 10,; 330 MPa and 380 MPa, respectively) and hardness (140 HV5 and 155 HV5, respectively) of the alloys. The solvus temperature of the  $\gamma'$  precipitates is less than the solution temperature of 1038 °C [2326], resulting in dissolution back into the  $\gamma$  matrix. The average hardness value for each sample obtained—is mostly due to the presence of small-sized  $\gamma'$  precipitates, which are precipitated during water quenching WQ, as shown in Fig. 4, and MC-type carbides and nitrides (NbC and TiN) that are precipitated), which precipitate randomly throughout the structure, some of which dissolve into the  $\gamma$  matrix during solution treatment.



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**Fig. 10**. Mean  $\gamma$ ' coarsening simulation of 945 (in black) and 945X (in red) STA samples <u>attern</u> two <u>step aging of at</u> (a) 718 °C and (b) 621 °C. The simulation (c) shows the effect of aging at 621 °C on mean  $\gamma$ ' coarsening after solution treatment.

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The secondary aging at 621 °C helps to achieve clear edge definition by fractionally increasing the  $\gamma$ ' size, which enhances the hardness and 0.2% strength of the alloys. The effect of single and double aging on the morphology of  $\gamma$ ' in ATI 718Plus has beenwas studied by using TEM-and reported in [2832]. It would be An interesting experiment would be to earry outapply TEM again onto both alloys, and to see the individual effects of each aging temperature after solution treatment. The  $\gamma$ ' coarsening simulation shown in Fig. 11-(c) demonstrates indicates the effect of individual aging at 621 °C for 8 h, showing an increase in the mean size of the  $\gamma$ ' particles to ~3 nm, which is contrary to the fractional increase of ~0.20 nm after consecutive double aging. The results of the theoretical simulations are close to the experimental observations (TEM) and could be used as a benchmark for comparison.

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